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Coordination Chemistry continues to be a basic presentation of this subject. It covers a host of interesting new areas in which the ideas of coordination chemistry have impact.

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Fred Basolo Evanston, Illinois March
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Fred Basolo was an American
inorganic chemist. He received his
Ph.D. at the University of Illinois at
Urbana-Champaign in 1943, under
Prof. John C. Bailar, Jr.. Basolo spent
his professional career at
Northwestern University. He was a
prolific contributor to the fields of

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coordination chemistry, organometallic, and bioinorganic chemistry, publishing over 400 papers. He supervised many Ph.D. students. With colleague Ralph Pearson, he co-authored the influential monograph "Mechanisms of Inorganic

Fred Basolo - Wikipedia

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Coordination Chemistry, a first year text book, had two editions - 1964 and 1984 - and was widely used in introductory chemistry courses for many years. In 2003, John Burmeister and Basolo edited a collection of his published articles for volume 13 in a series on well known chemist titled, On Being Well-Coordinated: A Half Century of Research .

From boyhood in the coal-mining village of Coello, Illinois, to winning the

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Priestly Medal and becoming the president of the American Chemical Society, Professor Emeritus Fred Basolo of Northwestern University traces the intertwined development of his life, career, and the field of inorganic chemistry. With over a hundred photographs and dozens of structures and equations, *From Coello to Inorganic Chemistry* details the major innovations, travels, family life, and guests hosted while helping to build one of the world's leading inorganic chemistry departments from its humble beginnings at Northwestern University. Students and chemists with interests in bioinorganic chemistry, catalysis, nanoscience, new materials research, and organometallics can follow the emergence of inorganic chemistry as a rival to organic chemistry through the

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accomplishments of one of its most influential pioneers.

Both elementary inorganic reaction chemistry and more advanced inorganic theories are presented in this one textbook, while showing the relationships between the two.

Over the last decades several researchers discovered that children, pupils and even young adults develop their own understanding of "how nature really works". These pre-concepts concerning combustion, gases or conservation of mass are brought into lectures and teachers have to diagnose and to reflect on them for better instruction. In addition, there are 'school-made misconceptions' concerning equilibrium, acid-base or redox

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reactions which originate from inappropriate curriculum and instruction materials. The primary goal of this monograph is to help teachers at universities, colleges and schools to diagnose and 'cure' the pre-concepts. In case of the school-made misconceptions it will help to prevent them from the very beginning through reflective teaching. The volume includes detailed descriptions of classroom experiments and structural models to cure and to prevent these misconceptions.

Previously by Angelici, this laboratory manual for an upper-level undergraduate or graduate course in inorganic synthesis has for many years been the standard in the field. In this newly revised third edition, the manual has been extensively updated

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to reflect new developments in inorganic chemistry. Twenty-three experiments are divided into five sections: solid state chemistry, main group chemistry, coordination chemistry, organometallic chemistry, and bioinorganic chemistry. The included experiments are safe, have been thoroughly tested to ensure reproducibility, are illustrative of modern issues in inorganic chemistry, and are capable of being performed in one or two laboratory periods of three or four hours. Because facilities vary from school to school, the authors have included a broad range of experiments to help provide a meaningful course in almost any academic setting. Each clearly written & illustrated experiment begins with an introduction that highlights the theme of the experiment, often including a

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discussion of a particular characterization method that will be used, followed by the experimental procedure, a set of problems, a listing of suggested Independent Studies, and literature references.

Coordination chemistry is the study of compounds formed between metal ions and other neutral or negatively charged molecules. This book offers a series of investigative inorganic laboratories approached through systematic coordination chemistry. It not only highlights the key fundamental components of the coordination chemistry field, it also exemplifies the historical development of concepts in the field. In order to graduate as a chemistry major that fills the requirements of the American Chemical Society, a student needs to

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John Basolo and John Cummins
take a laboratory course in inorganic chemistry. Most professors who teach and inorganic chemistry laboratory prefer to emphasize coordination chemistry rather than attempting to cover all aspects of inorganic chemistry; because it keeps the students focused on a cohesive part of inorganic chemistry, which has applications in medicine, the environment, molecular biology, organic synthesis, and inorganic materials.

This profusely illustrated book, by a world-renowned chemist and award-winning chemistry teacher, provides science students with an introduction to atomic and molecular structure and bonding. (This is a reprint of a book first published by Benjamin/Cummings, 1973.)

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Explains the basics of inorganic chemistry with a primary emphasis on facts; then uses the student's growing factual knowledge as a foundation for discussing the important principles of periodicity in structure, bonding and reactivity. New to this updated edition: improved treatment of atomic orbitals and properties such as electronegativity, novel approaches to the depiction of ionic structures, nomenclature for transition metal compounds, quantitative approaches to acid-base chemistry, Wade's rules for boranes and carboranes, the chemistry of major new classes of substances including fullerenes and silenes plus a chapter on the inorganic solid state.

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